CHEMICAL INDUSTRY WASTEWATER TREATMENT

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Abstract

 Treatment of chemical industrial wastewater from building and construction chemicals factory and plastic shoes manufacturing factory was investigated. The two factories discharge their wastewater into the public sewerage network. The results showed the wastewater discharged from the building and construction chemicals factory was highly contaminated with organic compounds. The average values of COD and BOD were 2912 and 150 mgO₂/l. Phenol concentration up to 0.3 mg/l was detected. Chemical treatment using lime aided with ferric chloride proved to be effective and produced an effluent characteristics in compliance with Egyptian permissible limits. With respect to the other factory, industrial wastewater was mixed with domestic wastewater in order to lower the organic load. The COD, BOD values after mixing reached 5239 and 2615 mgO₂/l. The average concentration of phenol was 0.5 mg/l. Biological treatment using activated sludge or rotating biological contactor (RBC) proved to be an effective treatment system in terms of producing an effluent characteristic within the permissible limits set by the law.

 Therefore, the characteristics of chemical industrial wastewater determine which treatment system to utilize. Based on laboratory results

engineering design of each treatment system was developed and cost estimate prepared.

Key words: chemical industry, wastewater, treatment, chemical, biological

Introduction

 The chemical industry is of importance in terms of its impact on the environment. The wastewaters from this industry are generally strong and may contain toxic pollutants.

 Chemical industrial wastes usually contain organic and inorganic matter in varying degrees of concentration. It contains acids, bases, toxic materials, and matter high in biological oxygen demand, color, and low in suspended solids. Many materials in the chemical industry are toxic, mutagenic, carcinogenic or simply hardly biodegradable. Surfactants, emulsifiers and petroleum hydrocarbons that are being used in chemical industry reduce performance efficiency of many treatment unit operations (EPA, 1998). The best strategy to clean highly contaminated and toxic industrial wastewater is in general to treat them at the source (Peringer, 1997) and sometimes by applying onsite treatment within the production lines with recycling of treated effluent (Hu et al., 1999). Since these wastes differ from domestic sewage in general characteristics, pretreatment is required to produce an equivalent effluent (Meric et al., 1999).

 In chemical industry, the high variability, stringent effluent permits, and extreme operating conditions define the practice of wastewater treatment (Bury et al., 2002). Hu et al. 1999 proposed concept to select the appropriate treatment process for chemical industrial wastewater based on molecular size and biodegradability of the pollutants.

 Chemical industrial wastewater can be treated by some biological oxidation methods such as trickling filters, rotating biological contactor (RBC), activated sludge, or lagoons (Nemerow, and Dasgupta, 1991; Jobbagy et al., 2000). Pollutants with a molecular size larger than 10,000- 20,000, can be treated by coagulation followed by sedimentation or flotation (Hu et al., 1999).Waste minimization in the production process in chemical industry is the first and most important step to avoid waste formation during the production (Carini, 1999; Alvarez et al., 2004). Because of the fluctuation in the strength and flow rate, Bury et al; 2002 applied dynamic simulation to chemical-industry wastewater treatment to manage and control the treatment plant.

 The main objective of the present study was to evaluate the use of alternative methods for the treatment of chemical industry wastewater.

Materials and Methods

 For this study two factories represent the chemical industry discharging their wastewater into the sewerage system were selected (Table 1). Composite samples from the different departments and the final effluents were collected. Physicochemical analyses were carried out according to the APHA (1998). Laboratory experiments have been carried out to recommend the appropriate treatment. Chemical coagulation precipitation and biological treatment via aerobic systems were investigated.

Table (1) Basic Information about the Selected Factories

Chemical treatment

 Chemical treatment was applied using lime aided with ferric chloride and lime aided with aluminum sulfate. The optimum pH and coagulant dose values which gave the best removal were determined using a jar test procedure. A continuous chemical treatment unit (Abou-Ella *et al*. 1995) was operated at the optimum pH and coagulant dose. A schematic diagram and specification of the treatment unit are given in Table 2 and Figure 1.

Item	Unit	Flash	Flocculation	Sedimentation
		mixing	Tank	
Dimension	cm	$10 \times 7 \times 5$	$15\times10\times30$	$40\times15\times25$
Volume	cm^3	350	4500	15000
Flow rate	liter/hour 5		5	5
Detention	minute	4.2	54	180
time				

Table 2 Specification of the continuous chemical treatment unit

Biological Treatment

 Biological treatment via activated sludge and rotating biological contactor was carried out.

Activated Sludge Treatment Unit

 Batch laboratory experiments were carried out using activated sludge process. Two liters Plexiglas laboratory columns were used. The wastewater was inoculated with activated sludge from plant treating domestic sewage. Daily the aeration was stopped to let the sludge settle then the supernatant was drained and the column was refilled again with the wastewater till considerable amount of adapted sludge was produced.

 To study the effect of aeration period on the activated sludge, several experiments were conducted. A fixed amount of sludge (3-4 g/l) was transferred to a different column to which the pretreated wastewater was added. A detention time ranging from one hour to twenty-four hours was examined. Dissolved oxygen concentration was adjusted to maintain a minimum concentration of 2 mgO₂/l. Characterization of the treated wastewater was carried out after 60 minutes settlement, sludge analysis was also carried out.

Rotating Biological contactor (RBC) Unit

 The aerobic unit was based on bio-film reactor followed by sedimentation tank, Figure 2 (Watanabe *et al*, 1995; Bader, 1988). Table 3 represents the geometric data of the experimental RBC system.

Figure (1) Schematic Diagram of Continuous Chemical Treatment unit

Figure (2) Schematic Diagram of the RBC Unit

Table (3) Geometric data of the experimental RBC

	\Box No. of stages	$\overline{4}$
\Box	Arrangement of	4×8
	discs	
\Box	Disc diameter	14
	(cm)	
\Box	Total discs surface	0.95
	area, (m^2)	
\Box	Basin's volume in	5.19
	liters	
	\Box % submersion	50%
\Box	Specific surface	183
	area m ² /m ³	
\Box	Rotation speed,	4
	(rpm)	
	\Box Hydraulic load	0.107
	$m^3/m^2/d$	

Results and Discussion

Case study 1: Building and Construction Chemicals Factory

 The factory produces special building chemicals; concrete add mixture, painting and coating materials and bitumen emulsion. The factory produces 11 -15 m³/d of wastewater. Analysis of the end-of-pipe showed that the wastewater was highly contaminated with nonbiodegradable and toxic organic matter. This is obvious from the average values of BOD (150 mgO₂/l) and COD (2912 mgO₂/l), (Table 4). The BOD/COD ratio was 6% in average. The analysis detected the presence of phenol with a concentration reaches 0.3 mg/l. The oil & grease ranged between 149 and 600 mg/l with an average value of 371 mg/l. Average value of total suspended solids concentration was 200 mg/l.

Biological Treatment

 Biological treatment of the end-of-pipe wastewater using activated sludge was carried out. Analysis of the wastewater indicated deficiency in the nitrogen and phosphorous concentration. Nitrogen and phosphorous salts were added to adjust their concentration to suffice for biological treatment. Characteristics of the treated effluent did not comply with the permissible limits.

Chemical Treatment

 Chemical treatment using lime aided with ferric chloride and lime aided with aluminum sulfate was carried out on a bench scale, first to get the best coagulant and the optimum dose and pH then, a continuous system was used.

Parameters	Units	Min	Ma	Average	Permissibl
		$\ddot{}$	X.		e Limits
pH		6.1	9.5	7.5	$6 - 9.5$
Chemical Oxygen	mg	187	392	2912	1100
demand	$O_2/1$	$\overline{0}$	$\overline{4}$		
Biological Oxygen	mg	210	570	150	600
Demand	$O_2/1$				
Total suspended solids	mg/1	157	519	200	800
Phosphorous	mg P/l	0.8	30	9	25
Organic Nitrogen	mg	9	25	19	100
	N_2/l				
Phenols	mg/1	0.06	0.3	0.1	0.05
Oil & Grease	mg/1	149	600	371	100

Table (4) Characteristics of wastewater from the end-of-pipe (Building and construction chemicals factory)

**Average of 7 samples*

Bench scale chemical treatment

 Table 5 shows the results of the chemical coagulation– sedimentation of the end-of-pipe using lime aided with ferric chloride and lime aided with aluminum sulfate. Significant removal of COD, TSS and Oil & Grease were achieved. The removal efficiency of COD, TSS and Oil & Grease were 94%, 81% and 91%, respectively using lime aided with ferric chloride. The settling properties of the sludge in case of lime aided with ferric chloride were better than in case of lime aided with aluminum sulfate.

Continuous chemical treatment

 Based on the bench scale results the wastewater was chemically treated with Lime aided with ferric chloride using continuous system. The specification of the treatment unit is listed in Table 2. The characteristics of finally treated effluent were compatible with legislation for discharging in public sewer system (Table 6).

Table (5) Average results of the chemical treatment using different coagulant

(Building and construction chemical factory)

**Average of 5 samples*

Table (6) Characteristics of the chemically treated wastewater (Building and construction chemical factory)

**Average of 6 samples*

Design and Economic Study of the Treatment System

 Based on the laboratory results a final chemical treatment process design was developed (Figure 3). Cost estimate of the treatment system indicated that the construction cost is LE 211000, while the running cost is LE 70200, (Table 7).

Figure (3) Schematic Diagram of the chemical treatment system (Building and construction chemicals factory)

(Building and construction chemicals factory)

Case study 2: Plastic Shoes Manufacturing Factory

 The second case study involved wastewater discharged from plastic shoes manufacturing factory. The manufacturing process involves raw material (polymers) melting unit, forming the pattern in special moulds transfer the shoes to paint unit where it is sprayed with special dyes and solvents. A field survey indicated that the major source of pollution was the painting department.

 Wastewater discharged from the painting department was characterized by the high contents of organic compounds (Table 8). The mean values of the chemical oxygen demand and the biological oxygen demand were 15441 and 7776 mg O_2/l , respectively. The average phenol concentration was 0.93mg/l. Thus the domestic wastewater was mixed with the industrial wastewater to achieve an end-of-pipe effluent of lower organic load. Also, addition of domestic wastewater compensates deficiency of nitrogen and phosphorous concentration in the industrial wastewater. Meric et al. 1999 recommended biological treatment for such kind of wastewater regarding dilution requirements and nitrogen and phosphorus supplement.

The average values of COD and BOD of the final effluent of the factory after mixing were 5239 and 2615 mgO₂/l respectively (Table 8) which still exceeds the discharging limits into the sewer system.

Chemical Treatment

 Chemical treatment of the final effluent was carried out using lime in combination with ferric chloride and Lime with aluminum sulfate, however the characteristics of the treated effluent still did not comply with the permissible limits set by the Egyptian Law. These results are in agreement with Meric et al., 1999 who mentioned that methods such as coagulation, flotation, were not applicable for strong wastewater from

polyester manufacturing industry due to the soluble nature of the pollutants.

Parameters	unit	Painting dept.		Final Eff.			Egyptian	
		Min	Ma		Avg Min	Ma	Avg	permissi
			X.			X.		ble
								limits
pH		5.6	7.6	6.5	6.8	7.8	7.2	$6 - 9.5$
Chemical	mg	102	204	154	212	677	523	1100
Oxygen demand	$O_2/1$	54	90	41	$\overline{4}$	5	9	
Biological	mg	578	105	777	105	352	261	600
Oxygen demand	$O_2/1$	$\overline{0}$	$00\,$	6	$\overline{0}$	$\overline{4}$	5	
Total suspended mg/l		830	192	143	192	105	506	800
solids			$\overline{0}$	$\mathbf{1}$		$\overline{4}$		
Phosphorous	mg	$\overline{2}$	18	6	12.8	20	15.5	25
	P/1							
Organic	mg	79	598	338	17.2 210		92	100
Nitrogen	N_2/l							
Phenols	mg/1	0.6	1.2	0.93	0.12	1.3	0.5	0.05
Oil & Grease	mg/1	126	571	377	28	543	218	100

Table (8) Characteristics of the wastewater discharged from plastic shoes manufacturing factory

Biological Treatment

 Aerobic biological treatment using activated sludge and RBC were carried out

Activated Sludge Treatment Unit

 The reactor was fed with the end-of-pipe wastewater and operated at a detention time ranging from one hour to twenty-four hours using a MLSS of 3 g/l. Analysis of the treated effluent indicated that the highest BOD removal was achieved at a retention time of 24 hours (Table 9). Average residual values of COD, BOD, TSS and Oil and Grease were 376 mgO₂/l, 131 mgO₂/l, 12 mg/l and 26 mg/l, respectively. These values are in agreement with the standards set by the Egyptian law for discharging treated wastewater into the sewerage system.

Table (9) Characteristics of the treated wastewater using activated

sludge

(Plastic shoes manufacturing factory)

Rotating biological contactor unit

 The RBC was fed continuously with the final effluent with an organic load of 7.8 kgBOD/ $m³$.d for 4 months. The results in Table 10 and Figure 4, showed that the average COD and BOD concentration values of the treated effluent were 474 mgO₂/l and 277 mgO₂/l, respectively. The average residual value of the suspended solids was 76 with a removal value 88%. The oil and grease percentage removal was 93% with a residual value of 16 mg/l. Characteristics of the treated effluent using the RBC were within the permissible limits. These results are in agreement with Hu et al., 1999 who reported that pollutants with a high biodegradability, i.e., a high value of BOD/COD ratio, can be effectively treated using biological treatment process

Table (10) Characteristics of the treated wastewater using RBC

(Plastic shoes manufacturing factory).

Figure (4) Characteristics of the treated wastewater using RBC (Plastic shoes manufacturing factory)

Design and Economic Study of the Treatment System

 Based on the laboratory results a final biological treatment process design via activated sludge or RBC was developed (Figures 5 and 6). Cost estimate for the activated sludge indicated that the construction system is LE 313000, while the running cost is LE 113500, (Table 11). The construction cost of the RBC is LE 308000, while the running cost is LE 60500, (Table 12). The RBC system is recommended because the management and operation of the system is easier and technically feasible by the low-skilled personnel.

Figure (5) Schematic diagram of the activated sludge treatment system (Plastic shoes manufacturing factory)

Table (11) Dimensions and cost estimate of activated sludge system

(Plastic shoes manufacturing)

Figure (6) Schematic diagram of the rotating biological contactor system (Plastic shoes manufacturing factory)

Table (12) Dimensions and cost estimate of rotating biological contactor (Plastic shoes manufacturing factory)

Conclusion

 Characteristics of chemical industrial wastewater determine the adequate treatment system.. More specifically; concentration, molecular size, solubility, toxicity and biodegradability of the pollutants.

 Dilution of chemical industrial pollutants using domestic sewage in the factory effectively decreases the concentration and toxicity of the pollutants and subsequently increases the efficiency of the biological treatment. Also dilution with sewage is cost effective since no chemical salts are required to provide nutrients.

References

- Abou-Elela, S.I., El-Kamah, E.M., Aly, H.I. and Abou-Taleb, E. (1995) Management of wastewater from the fertilizer industry. Wat. Sci. Tech. Vol. 32, No. 11, pp. 45-54.
- Alvarez, D., Garrido, N. Sans, R. and Carreras, I. (2004) Minimizationoptimization of water use in the process of cleaning reactors and containers in a chemical industry. Journal of Cleaner Production, 12, pp. 781-787
- American Public health Association, (1998) Slandered Methods for the Examination of Water and Wastewater. 20th ed., Washington. DC.
- Badr, N. M. (1988) Factors affecting nitrification of wastewater in the rotating biological contactor, M.Sc. thesis. Cairo University, faculty of science, Cairo, Egypt.
- Bury, S.J., Groot, C.K., Huth, C. and Hardt, N. (2002). Dynamic simulation of chemical industry wastewater treatment plants. Water Science & Technology, Vol. 45 No 4-5 pp 355–363
- Carini, D. (1999) Treatment of industrial wastewater using chemicalbiological sequencing batch biofilm reactor (SBBR) processes.

Ph.D. thesis Swiss Federal Institute of Technology, Zurich, Switzerland.

- EPA, (1998). Wastewater Treatment Technologies in: Pollution Prevention (P2) Guidance Manual for the Pesticide Formulating, Packaging and Repackaging Industry including implementing the P2 alternative, EPA, 821-B-98-017 June 1998, Pp. 41-46
- Hu, H.-Y., Goto, N. and Fujie, K. (1999). Concepts and methodologies to minimize pollutant discharge for zero-emission production. Wat. Sci. Tech. Vol. 10-11, Pp. 9-16.
- Jobbagy, A., Nerbert, N., Altermatt, R.H. and Samhaber, W.M. (2000) Encouraging filament growth in an activated sludge treatment plant of the chemical industry. Wat. Res. Vol. 34, No. 2, pp. 699-703.
- Meric, S., Kabdash, I., Tunay, O. and Orhon, D. (1999). Treatability of strong wastewaters from polyester manufacturing industry. Wat. Sci. Tech., Vol. 39, No. 10-11, Pp. 1-7
- Nemerow, N.L. and Dasgupta, A. (1991) "Industrial and Hazardous waste Treatment "Van Nostrand Reinhold, New York.
- Peringer, P. (1997). "Biologischer Abbau von Xenobiotica." BioWorld (1), pp. 4-7.
- Watanabe, Y., Okabe, S., Hirate, K. and Masuda, S. (1995) "Simultaneous Removal of Organic Material and Nitrogen by Micro-Aerobic Biofilms". Wat. Sci. Tech., Vol. 31, No. 1, pp. 195- 203.