





STUDY ON NATURE/SYNTHESIZED POLYMERS COMBINED WITH NANOPARTICLES ON SECONDARY RECOVERY TO MAXIMIZE OIL RECOVERY

Presented By:

Professor Attia Mahmoud Attia

Eng. Mohamed Sadek Hamedi

Co-Authors

Eng. Zeyad Tamer Alshony

Eng. Waleed Aly Abdelsabour

المنظمة العربية للتنمية الإدارية - جامعة الدول العربية



Outlines

- 1 Introduction
- 2 Problem statement
- 3 Research Objectives
- 4 Experimental Setup
- 5 Research Methodology
- 6 Results
- 7 Conclusion
- 8 References











Introduction

- Enhanced Oil Recovery (EOR) techniques have been employed to increase the recovery factor of oil reservoirs.
- Polymer flooding is a well-established technique for enhancing oil recovery from reservoirs.
- The process involves injecting water-soluble polymers into the reservoir to increase the viscosity of the injected fluid, reduce the mobility ratio, and improve sweep efficiency.







Introduction

- Several polymers have been used for this purpose, including Xanthan Gum, and Partially Hydrolyzed Polyacrylamide (HPAM).
- Determination of the optimum polymer concentration is one of the crucial steps in the planning phase of a certain polymer flooding project.
- Most of polymers used in EOR projects loses their abilities of viscosifying and degrade at harsh reservoir conditions (HSHT).







Problem Statement

 High salinities can accelerate the degradation of polymers. <u>Salts</u> present in the reservoir brine can lead to the breaking of polymer chains, reducing their viscosity and effectiveness as flow modifiers. This degradation can result in reduced oil recovery efficiency.







Study Objectives

- This study aims to give a comparison between Xanthan Gum (natural) and HPAM (synthesized) under various conditions and their effect on oil recovery.
- Observe the effect of silicon dioxide SiO₂ (nanoparticles) on the oil recovery, Use Silicon dioxide combined with each of the polymers and observe their effects on the oil recovery.





SCHE

Factor 1 A:XG conc ppm Response 1 Oil Recovery % Res m 500 1 1625 1 1002.5 1 1250 1 747.5 2000	Total Total Total Total 10 10 0.0 44473866 0.0 0.0 10 10 0.000088 0.240003860 0.0000850 0.240003860 0.0000850 0.240003860 0.0000850 0.240003860 0.00008500 0.0000850 0.00008500	Factor 1 A:XG conc ppm Response 1 Oil Recovery % Res m 500 0.607143 1625 0.611607 1250 0.691071 1002.5 0.671429 1250 0.691071 1250 0.691071 1250 0.691071 1250 0.691071 1250 0.691071 2000 0.601221		100 100 300 200 A15 sec = 130 100 0 600327 0.001971 0 1600009 = 0.0029 0.001971 0 1600009 = 0.0029 101	
Creating design:	Collecting data:	Data Entry:	Analysis and	Optimizing Model:	Visualization
Using statistical	According to the	The data	Modelling:	The software offers	and
analysis, it	generated design	collected is	It builds a model	tools to optimize the	Interpretation:
generates a design	matrix, the actual	recorded in the	to represent the	system being	It provides
matrix.	experiment is	design matrix.	relationship	studied to identify	visualizations
	performed and	L]	between the	the optimal factor	and graphical
	then the collecting		factors entered	settings that	tools, such as
	of the response		and the responses	maximize the	3D plots,
	data occurs.		using ANN by	desired response	interaction
			analyzing the data.	which in our case	plots, and
الإدارية - جامعة الدول العربية	المنظمة العربية للتنمية			the oil recovery.	contour plots.







Methodology 1. Chemicals used:

 Table 1. Oil properties (lab measured)

	Acid number	0.8 mg KOH/gm oil
Oil Properties	Viscosity	2.464 cp at 30°C and 1 atm
on Properties	Density	0.814 gm/cc at 30°C and 1 atm
	API	41.81°

Table 2. Polymers and nanoparticles concentrations

	Туре			Col	ncentr	ations		
Polymers	HPAM (ppm)	500	747.5	10	02.5	1250	0 1625	2000
,	Xanthan Gum (ppm)	500	747.5	10	02.5	125	0 1625	2000
Nanoparticles	Silicon Dioxide SiO ₂ wt%	0.02475	5 0.0499	962	0.0	75	0.1125	0.15





5

Methodology

3



Table 3. Properties of the sand pack model, brine, formation water salinity used

Properties, unit	Values
Sand size, mm	$600 \ \mu.m \le \text{sand size} < 700 \ \mu.m$
Length, cm	27.5
Diameter, cm	5
Area, cm ²	19.635
Bulk volume, cc	587
Pore volume, cc	160
Porosity, %	27.2572402
Permeability, mD	688.697398
Formation Water Salinity	150,000
Injected Brine Salinity	35,000

Figure 1. schematic diagram of the displacement apparatus

4

1.

2

3.

4

5

Compressor

Sand pack

Graduated tube

Pressure regulator

Chemical reservoir







Methodology 3. Solution preparation

1. Design Matrix Generation: Use Design Expert software to generate chemical concentrations for flooding.

2. Polymer Preparation

- a. Weigh the required amount of polymer using a high-precision sensitive balance.
- b. Add the polymer to saline water with 35,000 ppm salinity (using NaCl).
- c. Mix the solution using a stirrer.
- **3. Preparation Timing**: Prepare chemical solutions right before flooding to avoid air exposure and precipitation.
- 4. Nano Solutions Handling: Keep nano solutions on the stirrer continuously during the flood to maintain

solubility. المنظمة العربية للتنمية الإدارية - جامعة الدول العربية







- 1. Sand Pack Saturation: Fully saturate the sand pack with brine.
- 2. Initial Permeability Assessment: Introduce brine to the saturated sand pack to assess permeability.
- 3. Reservoir Initiation
 - a. Inject oil to displace the brine.
 - b. Collect the oil effluent to determine the amount of oil displacing the brine.

4. Water Flooding Process

- a. Reintroduce brine to displace the oil in the sand pack model.
- b. Collect new oil samples over time, while maintaining pressure and flow rate.

5. Repeated Steps for Different Fluids

a. Repeat the process for polymer, nano, and combined floods.

b. Change only the displacing fluid in the oil displacement phase for each repetition. المنظمة العربية للتنمية الإدارية - جامعة الدول العربية







Methodology

5. The interfacial tension measurement (IFT)

The IFT calculations were carried out using EZTensiometer surface tension calculation software by entering the maximum balance reading obtained from the EZTensiometer by ROD device and recording the interfacial tension measurement.





1. IFT measurement:





Table 4. Measurements of the IFT of polymers and nano used.

	Chemical concentration	IFT mN/m
	500 ppm	28.151
m	747.5 ppm	24.65
m G	1002.5 ppm	17.671
Xanthan Gum	1250 ppm	22.263
Xa	1625 ppm	19.607
	2000 ppm	17.353
	500 ppm	75.342
	747.5 ppm	41.465
HPAM	1002.5 ppm	40
	1250 ppm	35.798
	1625 ppm	31.746
	2000 ppm	23.279
-	0.02475 wt%	28.385
Nano silica (Silicon Dioxide SiO ₂)	0.049969196778602 wt%	14.564
	0.075 wt%	10.789
Nano s Dio:	0.1125 wt%	4.82939
	0.15 wt%	19.0236

المنظمة العربية للتنمية الإدارية - جامعة الدول العربية

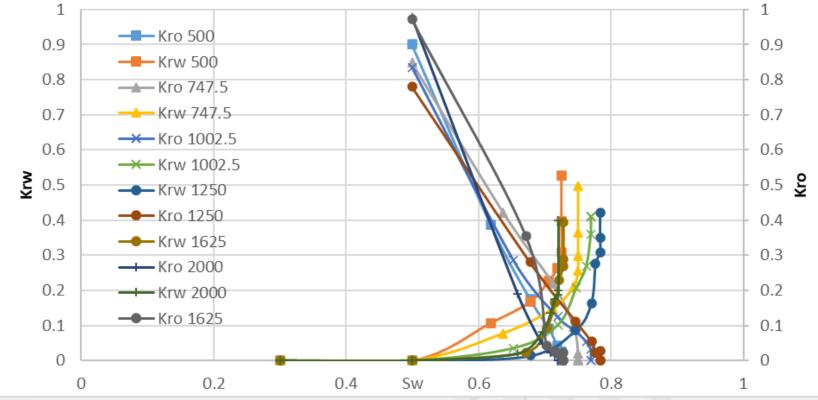






2. Wettability determination

2.1. Xanthan gum polymer flooding wettability



المنظمة العربية للتنمية الإدارية - جامعة الدول العربية

Figure 2. Relative permeability saturation curve for Xanthan-gum

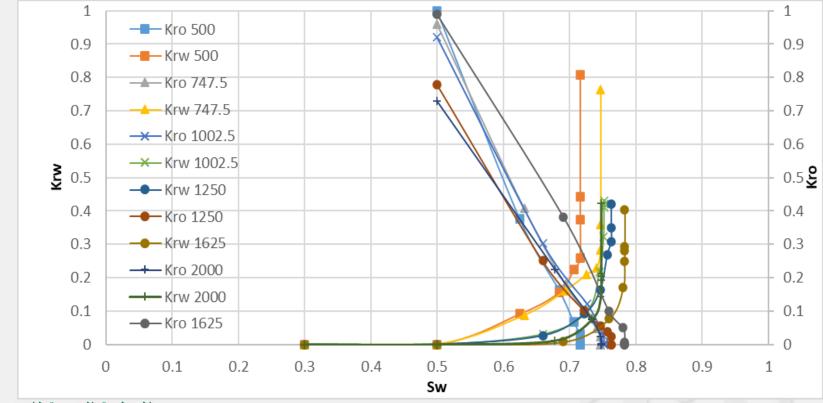






2. Wettability determination





المنظمة العربية للتنمية الإدارية - جامعة الدول العربية

Figure 3. Relative permeability saturation curve for HPAM

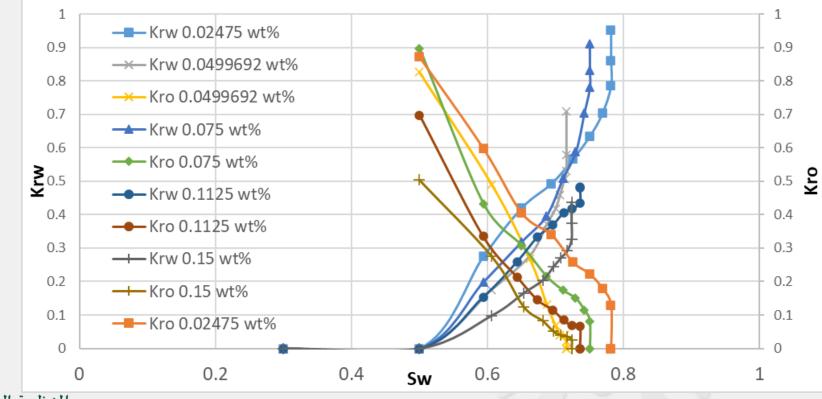






2. Wettability determination

2.3. Silicon dioxide nano flooding wettability



المنظمة العربية للتنمية الإدارية - جامعة الدول العربية

Figure 4. Relative permeability saturation curve for Silicon dioxide

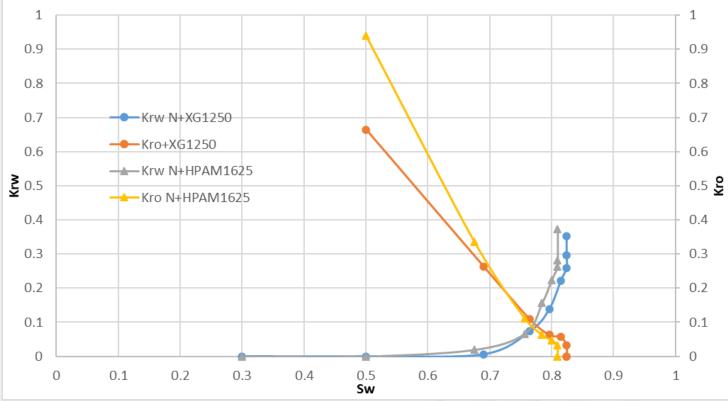






2. Wettability determination

2.4. Nanosilica-polymer combined flooding wettability



المنظمة العربية للتنمية الإدارية - جامعة الدول العربية

Figure 4. Relative permeability saturation curve for Nanosilica-polymer combined







Table 5. Summary of Swi @ Krw=Kro

Table 5. Summary of Swi @ Kiw-Kio				
	Chemical concentration	S _{wi} @ K _{rw} =K _{ro}		
Brine with	0.597			
	500 ppm	0.679		
<u> </u>	747.5 ppm	0.727		
G c	1002.5 ppm	0.728		
tha	1250 ppm	0.751		
Xanthan Gum	1625 ppm	0.681		
<u>^</u>	2000 ppm	0.69		
	500 ppm	0.681		
	747.5 ppm	0.699		
HPAM	1002.5 ppm	0.7265		
H	1250 ppm	0.727		
	1625 ppm	0.746		
	2000 ppm	0.732		
е (⁻ с	0.02475 Wt%	0.641		
Nano silica (Silicon Dioxide SiO ₂)	0.075 Wt%	0.65		
illic ide	0.1125 Wt%	0.612		
Nar (S)	0.15 Wt%	0.645		
	0.049969196778602 Wt%	0.66		
SiO ₂ (0.02475	<mark>0.778</mark>			
SiO ₂ (0.02475 Wt%) + HPAM1625		0.769		

المنظمة العربية للتنمية الإدارية - جامعة الدول العربية





WWW.ENSEG.ORG

	Chemical concentration	Recovered Oil (%)	
Brine with	46.786		
	500 ppm	60.714	
n	747.5 ppm	64.375	
U C	1002.5 ppm	67.143	
tha	1250 ppm	69.107	
Kanthan Gum	1625 ppm	61.1607	
×	2000 ppm	60.122	
	500 ppm	59.4643	
_	747.5 ppm	63.83928571	
HPAM	1002.5 ppm	64.55357143	
HP.	1250 ppm	66.16071429	
	1625 ppm	69.01785714	
	2000 ppm	63.88098214	
5	0.02475 Wt%	68.75	
siO siO	0.075 Wt%	64.2857143	
lico de d	0.1125 Wt%	62.3214286	
Nano silica (Silicon Dioxide SiO ₂)	0.15 Wt%	60.5357143	
	0.049969196778602 Wt%	59.4642857	
<mark>SiO2 (0.02475 Wt%) +XG1250</mark>		<mark>74.875</mark>	
SiO ₂ (0.02475 W	72.76785714		

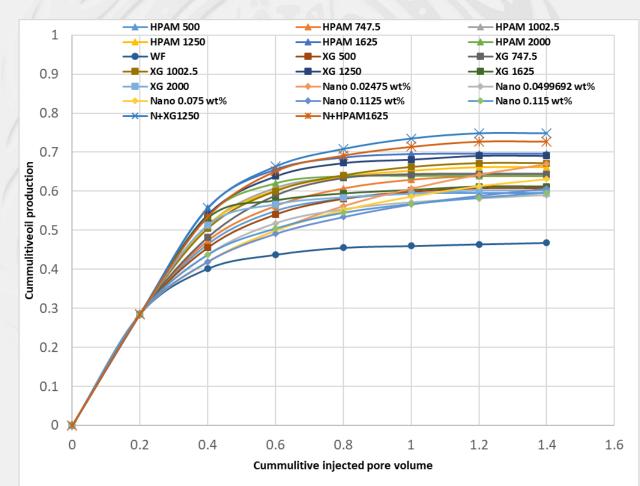


Figure 5. Relative permeability saturation curve for Nanosilicapolymer combined







- Xanthan gum was found to be the best polymer to be used as it gave the highest oil recovery at 1250 ppm.
- By adding the optimum nano concentration to the optimum polymer concentrations, the wettability was shifted from 0.746 in the HPAM flood to 0.769 and in the XG from 0.751 to 0.778 which led to a significant increase in the oil recovery.
- The oil recovery increases with the increase of the Xanthan-Gum concentration until it reached its peak at XG-concentration of 1250 ppm (69.107%) and for HPAM it reached its peak at the HPAM concentration of 1625 ppm (69.018%) and started to decrease again due to the adsorption effect.
- The oil recovery was at its best when combining the optimum nano-silica concentration (0.02475 wt%) with the optimum Xanthan-gum concentration (1250 ppm) and the optimum HPAM concentration (1625 ppm) to maximize the oil recovery (74.875%).









1. Agi, A., Junin, R., Abdullah, M.O., Jaafar, M.Z., Arsad, A., Sulaiman, W.R.W., Norddin, M.N.A.M., Abdurrahman, M., Abbas, A., Gbadamosi, A., 2020. Application of polymeric nanofluid in enhancing oil recovery at reservoir condition. J Pet Sci Eng 194, 107476.

2. Al-Shakry, B., Shiran, B.S., Skauge, T., Skauge, A., 2018. Enhanced oil recovery by polymer flooding: Optimizing polymer injectivity, in: Spe Kingdom of Saudi Arabia Annual Technical Symposium and Exhibition. OnePetro.

3. Attia, A., Canal, S.U., 2007. Relative Permeability and Wettability of Rocks Obtained From Their Capillary Pressure and Electrical Resistivity Measurements.

4. Attia, A.M., Musa, H., 2015. Effect of Sodium Magnesium Silicate Nanoparticles on Rheology of Xanthan Gum Polymer. Int J Sci Eng Res 6.

5. Azmi, G.E., Saada, A.M., Shokir, E.M., El-Deab, M.S., Attia, A.M., Omar, W.A.E., 2022. Adsorption of the Xanthan Gum Polymer and Sodium Dodecylbenzenesulfonate Surfactant in Sandstone Reservoirs: Experimental and Density Function Theory Studies. ACS Omega 7, 37237–37247. https://doi.org/10.1021/acsomega.2c03488

6. Castro-Garcia, R.-H., Maya-Toro, G.-A., Jimenez-Diaz, R., Quintero-Perez, H.-I., Díaz-Guardia, V.-M., Colmenares-Vargas, K.-M., Palma-Bustamante, J.-M., Delgadillo-Aya, C.-L., Pérez-Romero, R.-A., 2016. Polymer flooding to improve volumetric sweep efficiency in waterflooding processes. CT&F-Ciencia, Tecnología y Futuro 6, 71–90.

7. Dano, J., Abdelrahman, S., Ali, M., 2019. Simulation Study on Polymer Flooding for Enhanced Oil Recovery: A Case Study. Mater Today Proc 19, 1507–1513.

8. Druetta, P., Picchioni, F., 2019. Polymer and nanoparticles flooding as a new method for Enhanced Oil Recovery. J Pet Sci Eng 177, 479–495.







References

9. Dukeran, R., Soroush, M., Alexander, D., Shahkarami, A., Boodlal, D., 2018. Polymer flooding application in trinidad heavy oil reservoirs, in: SPE Trinidad and Tobago Section Energy Resources Conference. OnePetro.

10. El-Hoshoudy, A.N., Gomaa, S., Hassan, A., Attia, A.M., 2019. EFFECTS OF ALKALINE/POLYMER/NANOFLUIDS ON OIL RECOVERY AT HARSH RESERVOIR CONDITIONS. Petroleum & Coal 61.

11. Elsaeed, S.M., Zaki, E.G., Omar, W.A.E., Ashraf Soliman, A., Attia, A.M., 2021. Guar gum-based hydrogels as potent green polymers for enhanced oil recovery in high-salinity reservoirs. ACS Omega 6, 23421–23431.

12. Gomaa, S., Soliman, A.A., Nasr, K., Emara, R., El-hoshoudy, A.N., Attia, A.M., 2022. Development of artificial neural network models to calculate the areal sweep efficiency for direct line, staggered line drive, five-spot, and nine-spot injection patterns. Fuel 317. https://doi.org/10.1016/j.fuel.2022.123564

13. Hazarika, K., Gogoi, S.B., Kumar, A., 2022. Polymer flooding and its effects on enhanced oil recovery special reference to Upper Assam Basin. Petroleum Research.

14. Juárez, J.L., Bertin, H., Omari, A., Romero, C., Bourdarot, G., Jouenne, S., Morel, D., Neillo, V., 2020. Polymer Injection for EOR: Influence of Mobility Ratio and Slug Size on Final Oil Recovery, in: SPE Europec. OnePetro.

15. Keykhosravi, A., Vanani, M.B., Aghayari, C., 2021. TiO2 nanoparticle-induced Xanthan Gum Polymer for EOR: Assessing the underlying mechanisms in oil-wet carbonates. J Pet Sci Eng 204, 108756.

16. Mahran, S., Attia, A., Saha, B., 2022. Synthesis of green thermo-responsive amphoteric terpolymer functionalized silica nanocomposite derived from waste vegetable oil triglycerides for enhanced oil recovery (EOR). J Clean Prod 380, 135024.

17. Mahran, S., Attia, A., Saha, B., 2018. A REVIEW ON POLYMER FLOODING IN ENHANCED OIL RECOVERY UNDER HARSH CONDITIONS.







References

18. Moussa, E.O., Attia, A.M., 2016. Optimum Polymer Concentration in EOR. IARJSET 3, 4–15. https://doi.org/10.17148/iarjset.2016.31002

19. Petro, P., Eng, C., An, E.-H., Sm, D., Am, A., n.d. Petroleum & Petrochemical Engineering Journal Synthesis and Evaluation of Xanthan-G-Poly (Acrylamide) Co-Polymer for Enhanced Oil Recovery Applications Synthesis and Evaluation of Xanthan-G-Poly (Acrylamide) Co-Polymer for Enhanced Oil Recovery Applications.

20. Sanches, K.K.M., Moreno, R.B.Z.L., 2015. Polymer flooding: study of factors influencing the oil recovery. IV Journeys in Multiphase Flows (JEM 2015).

21. Sayyouh, M.H., Al-Blehed, M.S., Attia, A.M., 1993. The Effect of Alkaline and Polymer Additives on Phase Behaviour of Surfactant-Oil-Brine System At High Salinity Conditions. Revue de l'Institut Français du Pétrole 48, 359–369. https://doi.org/10.2516/ogst:1993023

22. Silveira, B.M.O., Lopes, L.F., Moreno, R., 2018. Polymer flooding in a high salinity heavy-oil reservoir. Brazilian Journal of Petroleum and Gas 12.

23. Soliman, A.A., El-Hoshoudy, A.N., Attia, A.M., 2020. Assessment of xanthan gum and xanthan-g-silica derivatives as chemical flooding agents and rock wettability modifiers. Oil & Gas Science and Technology–Revue d'IFP Energies nouvelles 75, 12.

24. Taborda, E.A., Franco, C.A., Lopera, S.H., Castro, R.H., Maya, G.A., Idrobo, E.A., Cortes, F.B., 2021. Effect of surface acidity of SiO2 nanoparticles on thermal stability of polymer solutions for application in EOR processes. J Pet Sci Eng 196, 107802.







FACULTY OF ENERGY AND ENVIRONMENTAL ENGINEERING

The authors acknowledge The Department of Petroleum Engineering and Gas Technology at Faculty of Energy and Environmental Engineering at The British University in Egypt (BUE), for funding this research and for providing the necessary facilities.

